Other Tests.—The substances were screened also for their actions on the isolated rabbit heart, following the method of Setuikar, *et al.*, ⁽¹⁾ changes in amplitude of contractions, rate of contractions, coronary flow, and resistance to anoxia¹² were recorded. Furthermore the substances were screened for their actions on blood pressure and respiration in rats anesthetized with 1.0 g/kg ip of urethan, on formaldehyde paw edema in rats, an electroshock convulsions in mice, and an CaCl₂-induced ventricular fibrillations in rats.

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(11) I. Selnikar, W. Murmann, and M. T. Ravasi, Arch. Intern. Pharmotrodyn. Therap., 131, 187 (1961).

(12) 1. Setnikar and M. T. Ravasi, *ibid.*, **124**, **1**16 (1960).

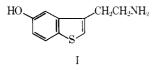
The Sulfur Analog of Serotonin¹

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A number of structural analogs of serotonin (5hydroxytryptamine) have been prepared.² Due to isoelectronic and steric relationships, the benzo[b]thiophene analog has been of particular interest as a possible agonist or antagonist of serotonin. The synthesis of this compound has proved to be refractory.³ We wish to report the synthesis of 3-(β aminoethyl)-5-hydroxybenzo[b]thiophene (I), and preliminary pharmacological evaluation.



3-Methyl-5-hydroxybenzo[b]thiophene,⁴ prepared by the cyclization procedure⁵ from *m*-hydroxyacetophenone, followed by decarboxylation of the resulting 3methyl-5-hydroxybenzo[b]thiophene-2-carboxylic acid, was converted to its benzoate ester. This ester was subsequently converted to 3-bromomethyl-5-benzoyloxybenzo[b]thiophene by the procedure of Chapman, *et al.*,⁶ and the corresponding carboxaldehyde was prepared in satisfactory yield *via* the Sommelet reaction,⁷ without hydrolysis of the ester linkage. The 5-benzoyloxybenzo[b]thiophene-3-carboxaldehyde was then condensed with nitromethane, employing ammonium acetate as catalyst. Two products were isolated, the major product being 5-benzoyloxy-3-(2-nitrovinyl)benzo[b]thiophene, and the minor product being 5-

(3) (a) S. T. Reid, Ph.D. Thesis, The University of Glasgow, 1960; (b)
(a) Brown, Ph.D. Thesis, The University of Glasgow, 1962; (c) E. S. Neiss
(b) D. Thesis, University of Glasgow, 1962; (c) E. S. Neiss

Ph.D. Thesis, Indiana University, 1964.
(4) A. Ricci, N. P. Bun-Hni, P. Jacquignon, and M. Dufong, J. Heterocyclic Chem., 2, 300 (1965).

(5) E. Campaigne and R. E. Cline, J. Oxy. Chem., 21, 39 (1956).

(6) N. W. Chapman, K. Clarke, and B. bldop, J. Chem. Soc., 774 (1965)

(17) S. J. Angyal, Om. Reactions, 8, 197 (1954).

hydroxy-3-(2-nitrovinyl)benzo[b]thiophene. 5-Benzoyloxy-3-(2-nitrovinyl)benzo[b]thiophene was reduced with lithium aluminum hydride and the reaction was worked-up according to the method of Martin-Smith, *et al.*^{2a} Compound I was isolated as the hydrochloride.

The central nervous system effect of 1 and 5-hydroxytryptophan (11) was studied by amplitude analysis of the cortical electroencephalogram (EEG) of male albino rabbits.⁸ It has been demonstrated that animals given intravenous doses of II have significant increases of brain scrotonin.^{9,19} Administration of 400 µg/kg of I or II resulted in desynchronization of the EEG, indicative of a highly stimulated state: I caused a drop of the mean energy content (MEC) to 32.5% below control levels and II a drop of 41.0%. No peripheral effects were observed in rabbits treated with I. Animals pretreated with pentobarbital (3 mg/kg) showed a very similar polyphasic response to both 1 and 11: at 50 $\mu g/kg$ both were synergistic to the sedative. maximally stimulated at 200 μ g/kg, and again sedative at 500 μ g/kg. The barbiturate effect was 50% reversed (RD₅₀) at a dose of 160 μ g/kg of I and 140 μ g/kg of II.

Further studies on the synthesis and biological activity of benzo|b|thiophene analogs of biologically active indole derivatives are currently under investigation.

Experimental Section 11

5-(α -Methyl-3-hydroxybenzylidene)rhodanine...-Rhodanine (67 g, 0.5 mole) was added to a solution of 4 g of ammonium acctate and 12 ml of glacial acetic acid in 400 ml of dry benzene and boiled for a few minutes. *m*-Hydroxyacetophenone (68 g, 0.5 mole) was added to the hot reaction mixture and the flask was connected to a Dean-Stark trap. The reaction mixture was refluxed vigorously until solid began to separate, cooled to roum temperature, and filtered. The yellow precipitate was washed with two 100-ml portions of water and air dried. Recrystallization from dioxane-water gave 100 g (80%) of product which melted at 201-202°. An analytical sample melted sharply at 207°.

Anal. Caled for C₁₁H₂NO₂S₂: S, 25.52. Found: S, 25.64.

β-Methyl-β-(3-hydroxyphenyl)-α-mercaptoacrylic Acid. 5-(α-Methyl-3-hydroxybenzylidine)rhodanine (50 g, 0.20 mole) was added to a stirred solution of 1 h of 10% NaOH at 60°. The amber solution was hea(ed to 80° and stirred for 1 hr prior to saturation with NaCl and filtration through a Norit pad. The solution was cooled to 10° and slowly poured into 400 ml of 6 N HCl which was saturated with NaCl and cooled to 10°. The yellow solid was collected and dried to yield 39 g (75%) of prodnet which melted 128-129° after recrystallization from propanol: λ_{max}^{SP} 3.00 (intermolecular 11 banded OII), 3 4 (OII of acid), 3.95 (very weak) (SH), 5.95 (C=O), 6.25 (aryl conjugated C=-C), 12.63, 14.2, and 14.45 μ (1,3-disnbstituted benzene).

.tual. Caled for C₁₀H₁₀O₃S: S, 15.25. Found: S, 15.11.

3-Methyl-5-hydroxybenzo[b]**thiophene-2-carboxylic** Acid. β -Methyl- β -(3-hydroxyphenyl)- α -mercaptoacrylic acid (20 g, 0.095 mole) and 30 g of 1; were allowed to gently reflexed for 15 hr in 500 ml of dry dioxane. The solution was reduced to half its volume under reduced pressure and poored into 2 1, of cold

⁽¹⁾ Part IX in the series of benza[5](biophene derivatives. For Part VIII see F. Campaigne and E. S. Neiss, J. Heterogarlie Chem., 3, 46 (1966). Taken from a thesis to be submitted by T. Bosin to Indiana University for the Ph.D. degree.

^{(2) (}a) J. J. Lewis, M. Martin-Smith, T. C. Muri, S. N. Nanjappa, and S. T. Reid, J. Med. Chem., 6, 711 (1963); (b) R. Foster, H. R. Ing, and E. F. Rogers, J. Chem. Soc., 1671 (1957); (c) C. Ainsworth, J. Am. Chem. Soc., 79, 5245 (1957); (d) G. Hallmann and K. Haegele, Ann., 662, 147 (1963).

⁽⁸⁾ L. Gubbstein and R. A. Bork, *Intern. Rev. Neurobiol.* 8, 205 (1965). While a direct comparison with semiopin will be more valuable, this preliminary screen doministrates the high activity of I in a test where scrotonio is essentially inactive.

⁽i) R. B. Brodie, E. G. Tuprich, R. Kuntzman, and P. A. Shore, J. Pharmacol., 119, 461 (1957).

⁽¹⁰⁾ S. Udenfriend, D. F. Wogdanski, and H. Weissbach, Federation Proc., 15, 493 (1956).

⁽¹⁾ Melting points were taken in a Mel-Temp capillary melting point apparatus and are uncorrected. The microanalyses were performed by Midwest Microlabs, Inc., Indiampolis, 1pd. Infrared spectra were determined with a Perkin-Elmer Mindel 157 Infrareod and altraviolet spectra were obtained with a Bansch and Lamb Spectrophy 505. Pur spectra were taken on a Varian A-60 spectrometer.

water containing 60 nil of saturated NaHSO₃ solution. The entire dilution was extracted with four 250-ml portions of ether and the combined ether extracts in turn were extracted with two 150-ml portions of 10% NaHCO₃. The dark alkaline solution was boiled with Norit, filtered, cooled, and acidified with 6 N HCl. The tan precipitate was collected and dried to yield 17.5 g (88%) of product which melted at 241–243°. After two recrystallizations from 95% ethanol, white crystals were obtained and melted at 254–255° dec; $\lambda_{\text{max}}^{\text{KBr}}$ 3.0 (OH), 3.4–4.1 (OH of CO₂H), 6.05 μ (C=O).

Anal. Caled for $C_{10}H_8O_3S$: C, 57.68; H, 3.90; S, 15.39. Found: C, 57.41; H, 3.96; S, 15.11.

3-Methyl-5-hydroxybenzo[b]**thiophene**.—3-Methyl-5-hydroxybenzo[b] thiopheue-2-carboxylic acid (4 g, 0.019 mole) was slowly heated during 1 hr to 210° in 40 ml of redistilled quinoline containing 2 g of Cu powder. The temperature was maintained at 210° for a further hour and then the reaction mixture was cooled, diluted with 150 ml of ether, and filtered. The etheral solution was extracted with 6 N HCl until acidic to congo red paper. The ether layer was then extracted with 20% NaOH, decolorized with Norit, and acidified with 6 N HCl. The acidic solution was extracted with three 75-ml portions of ether and dried (Na₂SO₄), and the ether was removed to yield a brown oil. The oil was dissolved in boiling cyclohexane and upon cooling gave 1.8 g (57%) of product. Analytical material was obtained after three recrystallizations from cyclohexane, mp 93–94°;⁴ $\lambda_{max}^{\text{KBy}}$ 2.96 (OH), 3.25 (=CH), 3.41 (CH₃), 8.22, 8.59, 7.20 μ (phenolic OH). The ultraviolet spectrum indicated $\lambda_{max}^{\text{Byg}}$ ethanoi in m μ (ϵ): 238 (21,800), 265 (5270), 270 sh (4660), 308 (3060), and 317 (2880); umr (CDCl₃), 8.22 (3 H, doublet), 5.72 (1 H, singlet), 6.7–7.2 (3 H, multiplet), 7.5–7.67 (1 H, doublet).

Anal. Caled for C₉H₈OS: S, 19.45. Found: S, 19.41.

The picric acid charge-transfer complex was prepared in the usual manner,¹² as tiny orange needles which melted at 150–151° after recrystallization from ethanol.

Anal. Caled for $C_{15}H_{11}N_3O_5S$: N, 10.68; S, 8.14. Found: N, 10.90; S, 8.46.

3-Methyl-5-benzoyloxybenzo[b]**thiophene**.—A solution of 2.76 g (0.017 mole) of 3-methyl-5-hydroxybenzo[b]thiophene, 20 ml of dry pyridine, and 2.36 g (0.017 mole) of benzoyl chloride was heated to gentle reflux for 3 hr. The reaction mixture was cooled to room temperature and poured into 125 ml of ice water. The solid which separated was collected and washed with 5% NaHCO₃ prior to recrystallization from ethanol to yield 3.85 g (85%) of white needles: mp 67–68.5°; $\lambda_{\rm mat}^{\rm KBr}$ 3.28 (=CH), 3.45 (CH₃), 5.79 (C=O), 6.27 μ (C=C aromatics); nmr (CDCl₃), δ 2.33 (3 H, doublet), 7.0–8.5 (9 H, multiplet).

Anal. Caled for $C_{16}H_{12}O_2S$: C, 71.62; H, 4.51; S, 11.95. Found: C, 71.87; H, 4.88; S, 11.82.

3-Bromomethyl-5-benzoyloxybenzo[b]**thiophene**.—A solution containing 0.5 g (1.87×10^{-3} mole) of 3-methyl-5-benzoyloxybenzo[b]thiophene and 0.019 g of benzoyl peroxide dissolved in 20 ml of reagent CCl₄ was heated to a gentle reflux whereupon 0.33 g (1.87×10^{-3} mole) of recrystallized N-bromosuccinimide was added and two 200-w lights focused on the reaction flask. The reaction was allowed to reflux for 2 hr, allowed to cool to room temperature, and filtered to remove succinimide. The CCl₄ was removed under a stream of N₂ to yield a yellow solid. The crude product was recrystallized from cyclohexane to give 0.51 g (80%) of white plates: mp 115-116°; $\lambda_{max}^{KBr} 3.28 (=CH)$, 5.79 (C=O), 6.26 μ (C=C aromatic); nmr (CDCl₃), δ 4.62 (2 H, singlet), 7.0-8.4 (9 H, multiplet).

Anal. Calcd for $C_{16}H_{11}B_1O_2S$: C, 55.34; H, 3.19; Br, 23.02. Found: C, 55.03; H, 3.25; Br, 23.59.

5-Benzoyloxybenzo[b]**thiophene-3-carboxaldehyde**.—A solution of 2.87 g (8.27×10^{-3} mole) of 3-bromomethyl-5-benzoyloxybenzo[b]thiophene and 1.18 g (8.40×10^{-3} mole) of hexamethylenetetramine in 25 ml of CHCl₃ was refluxed for 6 hr, after which time it was cooled to room temperature and the CHCl₃ was removed under reduced pressure leaving a light tan crude hexamine salt. This salt was treated with 30 ml of 50% aqueous acetic acid and the resulting solution was heated to reflux for 3 hr. At the completion of the heating period, 40 ml of water and 7 ml of concentrated HCl was added and the mixture refluxed for an additional 5 min. The reaction mixture was allowed to stand overnight, then diluted with 200 ml of water and ex-

(12) R. Shriner, R. Fuson, and D. Chrtin, "The Systematic Identification of Organic Components," John Wiley and Sons, Inc., New York, N. Y., 1948. Anal. Caled for $C_{16}H_{10}O_3S$: C, 68.06; H, 3.57; S, 11.36. Found: C, 67.87; H, 3.55; S, 11.06.

The semicarbazone was prepared by the usual procedure¹² as white plates, mp $224-225^{\circ}$, following recrystallization from ethanol.

Anal. Calcd for $C_{17}H_{18}N_3O_3S;\ C,\ 60.16;\ H,\ 3.86;\ N,\ 12.38.$ Found: C, 59.99; H, 3.82; N, 12.59.

5-Benzoyloxy-3-(2-nitrovinyl)benzo[b]**thiophene**.—A solution of 0.3 g (1.1 × 10⁻³ mole) of 5-benzoyloxybenzo[b]thiophene-3carboxaldehyde and 0.12 g of ammonium acetate in 6 ml of nitromethane was brought to a gentle reflux and maintained for 1 hr, after which time the excess nitromethane was removed under a stream of N₂ to leave a yellow solid. The crude solid was dissolved in boiling benzene, filtered, and upon cooling gave 0.275 g (80%) of yellow needles: mp 179–180°; λ_{max}^{KB} 3.27 (=CH), 5.79 (C=O), 6.14 (C=C olefin), 6.64 and 7.5 μ (NO₂).

Anal. Caled for $C_{17}H_{11}NO_4S$: C, 62.76; H, 3.41; N, 4.31. Found: C, 63.20; H, 3.60; N, 4.23.

5-Hydroxy-3-(2-nitrovinyl)benzo[b]thiophene.—This benzeneinsoluble material can be isolated from the condensation reaction between nitromethane and 5-benzoyloxybenzo[b]thiophene-3carboxaldehyde, and composes 5% of the crude reaction mixture. Recrystallization from chloroform gave gold needles: mp 227– 228° dec: $\lambda_{max}^{\rm EB}$ 3.0 (OH), 3.28 (=CH), 6.18 (C=C olefin), 6.70 and 7.6 μ (NO₂).

Anal. Caled for $C_{10}H_7NO_3S$: C, 54.29; H, 3.19; N, 6.33. Found: C, 54.41; H, 3.32; N, 6.48.

3-(β -Aminoethyl)-5-hydroxybenzo[b]thiophene Hydrochloride. —Lithium aluminum hydride (4.0 g, 0.105 mole) was added to 100 ml of dry tetrahydrofuran (THF), followed by the dropwise addition of 2.25 g (6.9 × 10⁻³ mole) of 5-benzoyloxy-3-(2-nitrovinyl)benzo[b]thiophene which was dissolved in 50 ml of dry THF. The reaction mixture was gently refluxed for 6 hr before the excess LiAlH₄ was decomposed by the careful addition of water, 200 ml of 2 N NaOH was added, and the entire reaction mixture was filtered. The THF was removed by distillation, and the basic solution was saturated with CO₂ (pH 8.3) and extracted continuously with ether for 48 hr. Dry HCl was passed into the ether solution, yielding a yellow oil which solidified under vacuum. Recrystallization from methanol-ethyl acetate gave 0.19 g (12%) of white plates: mp 195-196.5°; $\chi_{max}^{Br} 3.00$ (OH), 3.23-3.40 (NH₃⁺), and strong absorptions at 6.84, 6.96, 7.25, 8.03, and 8.19 μ . The ultraviolet spectrum indicated χ_{max}^{brg} sthand in m μ (ϵ): 237 (18,630), 264 (5520), 270 sh (4820), 309 (3310), and 316 (3200).

Anal. Caled for $C_{10}H_{12}$ CINOS: C, 52.28; H, 5.27; N, 6.09. Found: C, 52.38; H, 5.45; N, 6.10.

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Some *p*-Hydroxyphenoxyacetic Acid Derivatives

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The antithyroid activity of α -methyl- β -(3,5-diiodo-4-hydroxyphenyl)propionic acid³ and of esters of 3,5-

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⁽²⁾ This investigation was supported by Grant AM-06480, National Institutes of Health.

⁽³⁾ S. B. Barker, H. B. Dirks, Jr., W. R. Garliek, and H. M. Kli(gaarl, Proc. Soc. Exptl. Biol. Med., 78, 840 (1951).